Mole Calculation Worksheet

- How many moles are in 15 grams of lithium? 1)
- 2) How many grams are in 2.4 moles of sulfur?
- 3) How many moles are in 22 grams of argon?
- 4) How many grams are in 88.1 moles of magnesium?
- 5) How many moles are in 2.3 grams of phosphorus?
- How many grams are in 11.9 moles of chromium? 6)
- 7) How many moles are in 9.8 grams of calcium?
- 8) How many grams are in 238 moles of arsenic?

What are the molecular weights of the following compounds?

- 9) NaOH 11) MgCl₂ 13) Mn₂Se₇ H_3PO_4 10) H_2O 12) 14) $(NH_4)_2SO_4$
- 15) How many grams are in 4.5 moles of sodium fluoride?
- How many moles are in 98.3 grams of aluminum hydroxide? 16)
- How many grams are in 0.02 moles of beryllium iodide? 17)
- How many moles are in 68 grams of copper (II) hydroxide? 18)
- 19) How many grams are in 3.3 moles of potassium sulfide?
- 20)
- How many moles are in 1.2×10^3 grams of ammonia? How many grams are in 2.3×10^4 moles of calcium phosphate? 21)
- How many moles are in 3.4×10^{-7} grams of silicon dioxide? 22)
- 23) How many grams are in 1.11 moles of manganese sulfate?

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- How many grams are in 88.1 moles of magnesium? 4)
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What are the molecular weights of the following compounds?

- 11) 9) NaOH $MgCl_2$ Mn_2Se_7 13) 10) 12) H₃PO₄ H_2O 14) $(NH_4)_2SO_4$
- How many grams are in 4.5 moles of sodium fluoride? 15)
- 16) How many moles are in 98.3 grams of aluminum hydroxide?
- How many grams are in 0.02 moles of beryllium iodide? 17)
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- 23) How many grams are in 1.11 moles of manganese sulfate?

Mole Calculation Worksheet – Answer Key

- 1) How many moles are in 15 grams of lithium? 2.0 moles
- 2) How many grams are in 2.4 moles of sulfur? **77 grams**
- 3) How many moles are in 22 grams of argon? **0.55 moles**
- 4) How many grams are in 88.1 moles of magnesium? 2140 grams
- 5) How many moles are in 2.3 grams of phosphorus? **0.074 moles**
- 6) How many grams are in 11.9 moles of chromium? 619 grams
- 7) How many moles are in 9.8 grams of calcium? **0.24 moles**
- 8) How many grams are in 238 moles of arsenic? **17,800 grams**

What are the molecular weights of the following compounds?

- 9) NaOH **40.1 grams** 12) H₃PO₄ **98.0 grams**
- 10) H₂O **18.0 grams** 13) Mn₂Se₇ **663.0 grams**
- 11) $MgCl_2$ **95.3 grams** 14) $(NH_4)_2SO_4$ **132.1 grams**
- 15) How many grams are in 4.5 moles of sodium fluoride, NaF? 190 grams
- How many moles are in 98.3 grams of aluminum hydroxide, Al(OH)₃?

 1.26 moles
- 17) How many grams are in 0.02 moles of beryllium iodide, BeI₂? 5 grams
- How many moles are in 68 grams of copper (II) hydroxide, Cu(OH)₂?

 0.70 moles
- 19) How many grams are in 3.3 moles of potassium sulfide, K₂S? 360 grams
- 20) How many moles are in 1.2 x 10³ grams of ammonia, NH₃? 70 moles
- How many grams are in 2.3 x 10⁻⁴ moles of calcium phosphate, Ca₃(PO₃)₂? **0.064 grams**
- 22) How many moles are in 3.4×10^{-7} grams of silicon dioxide, SiO₂? 5.66 x 10⁻⁹ moles
- How many grams are in 1.11 moles of manganese sulfate, MnSO₄? **168 grams**