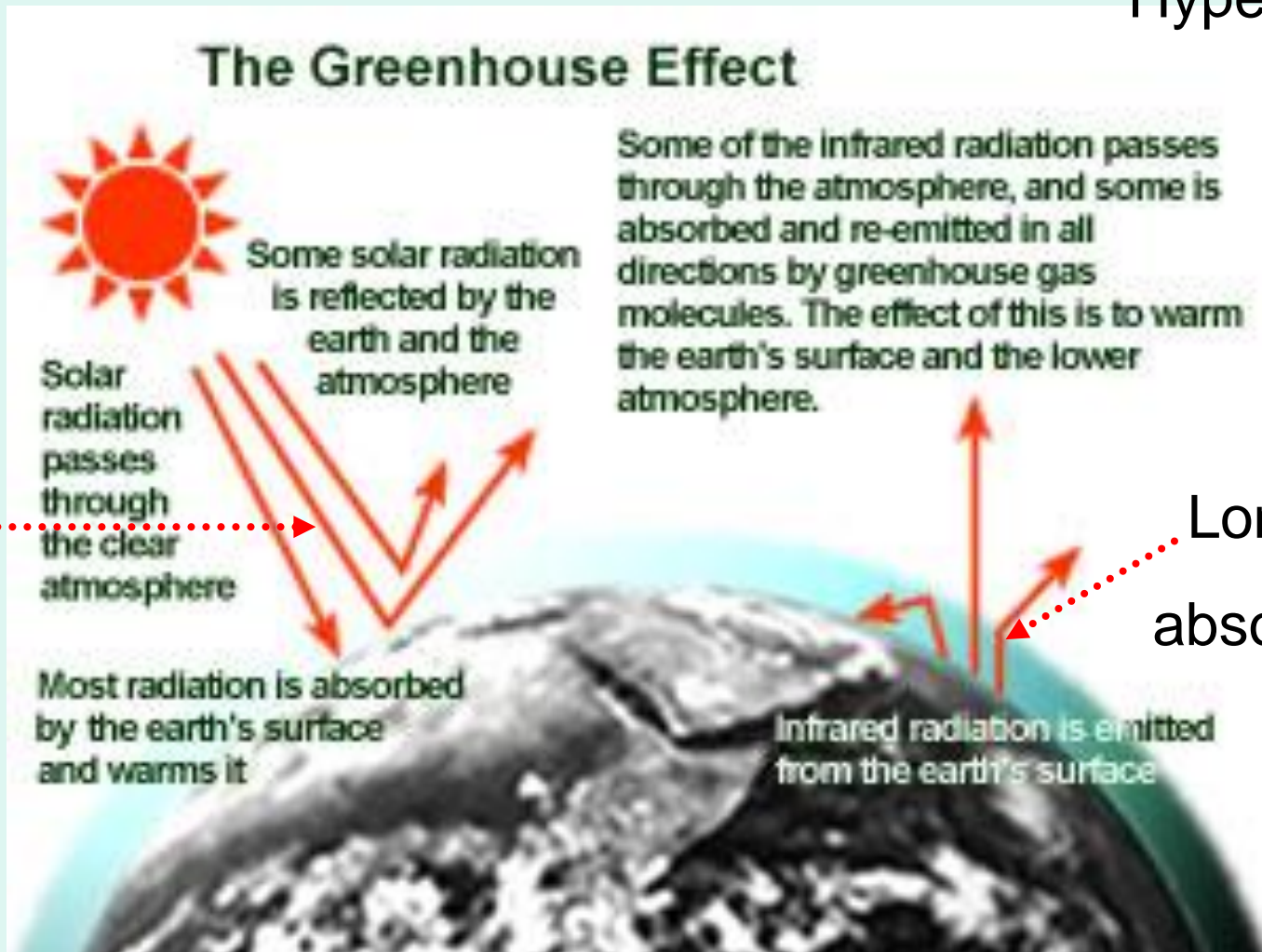


Greenhouse effect

Hyperlink



Short λ

not
absorbed

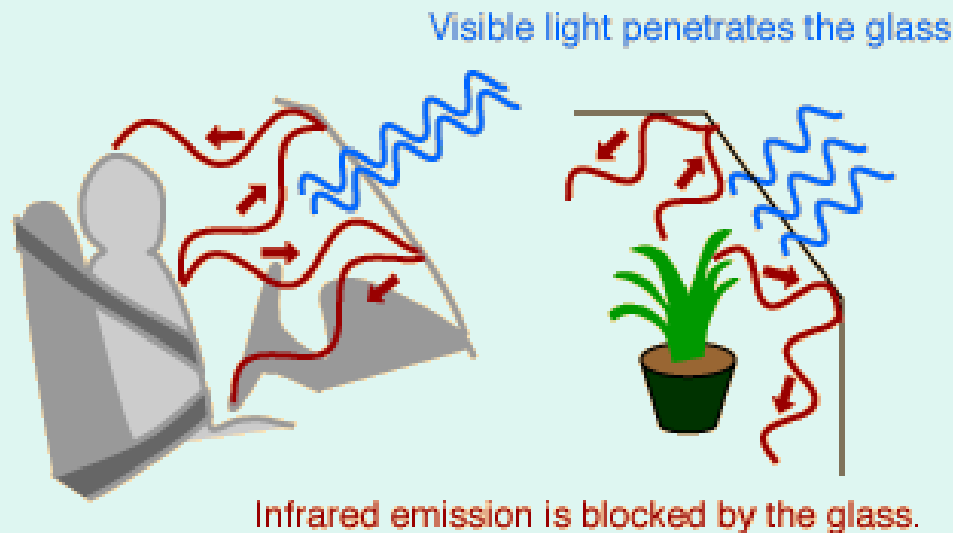
Long λ

absorbed

Solar constant

- The sun radiates $3.9 \times 10^{26} \text{W}$
- The Earth is a distance of $1.5 \times 10^{11} \text{m}$ from the sun
- Calculate the power per m^2 reaching the Earth.

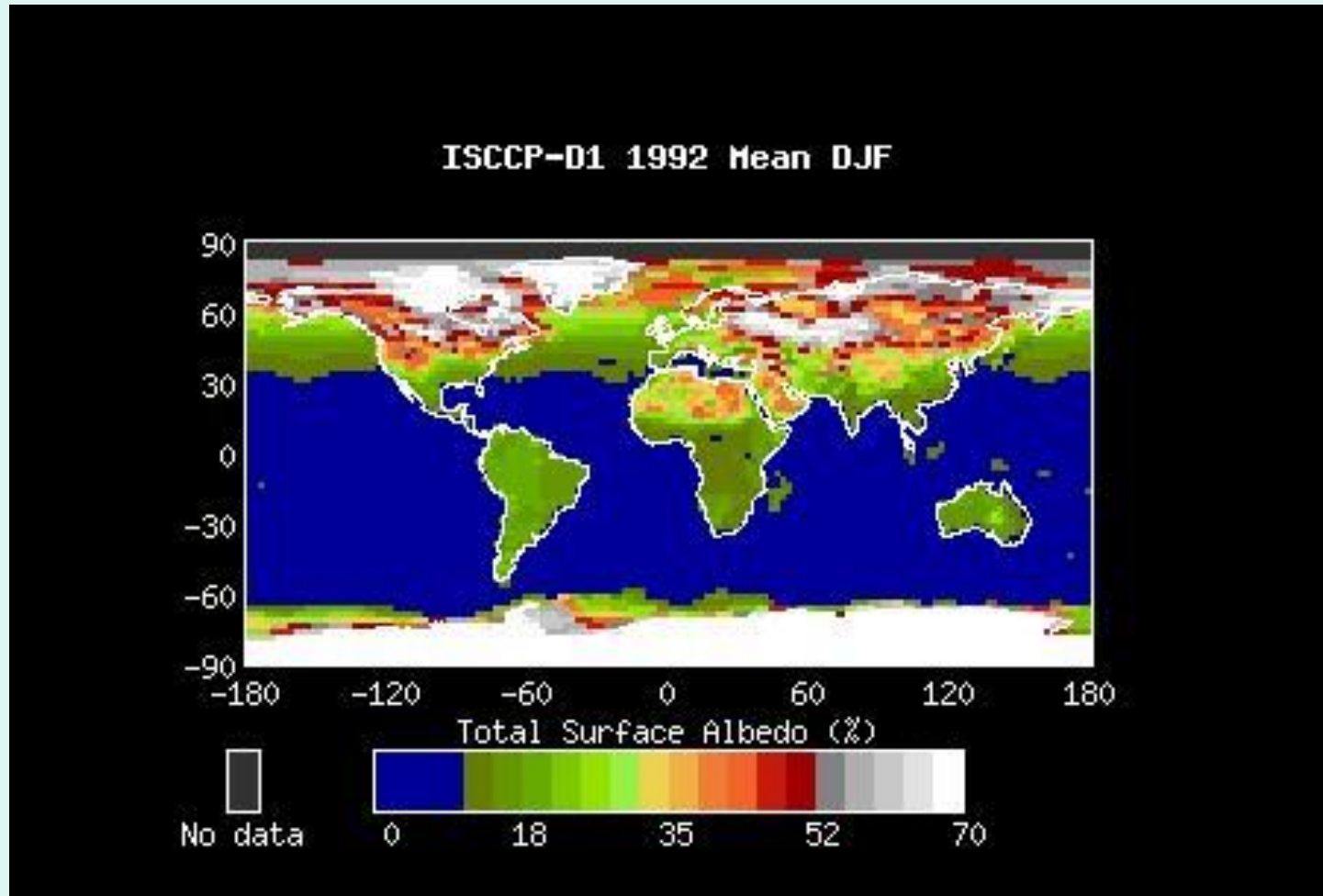
$$I = \frac{\text{power}}{A}$$



When the energy reaches the
Earth, what happens to it?

Albedo

$$\text{albedo} = \frac{\text{total scattered power}}{\text{total incident power}}$$



the fraction of the incident sunlight that is reflected

Variations in albedo

Sample albedos

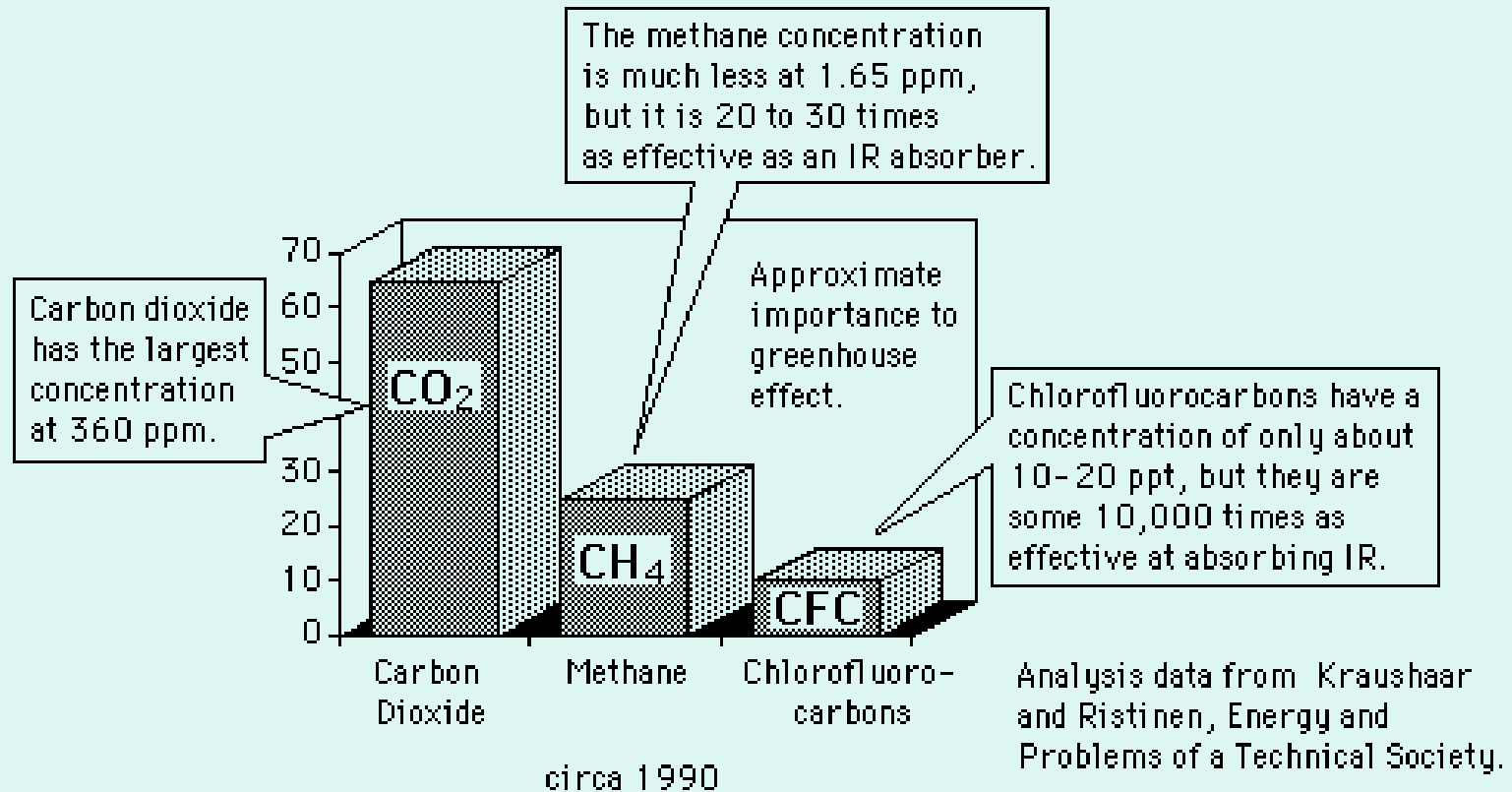
Surface	Typical Albedo
Fresh asphalt	0.04
Conifer forest (Summer)	0.08, 0.09 to 0.15
Worn asphalt	0.12
Deciduous trees	0.15 to 0.18
Bare soil	0.17
Green grass	0.25
Desert sand	0.40
New concrete	0.55
Fresh snow	0.80–0.90

The albedo also varies with factors like season, latitude and cloud cover

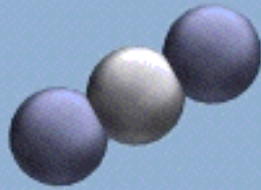
The average value on Earth is 0.3

Why does the reflected radiation
not escape into space?

Greenhouse gases



Absorption of IR radiation

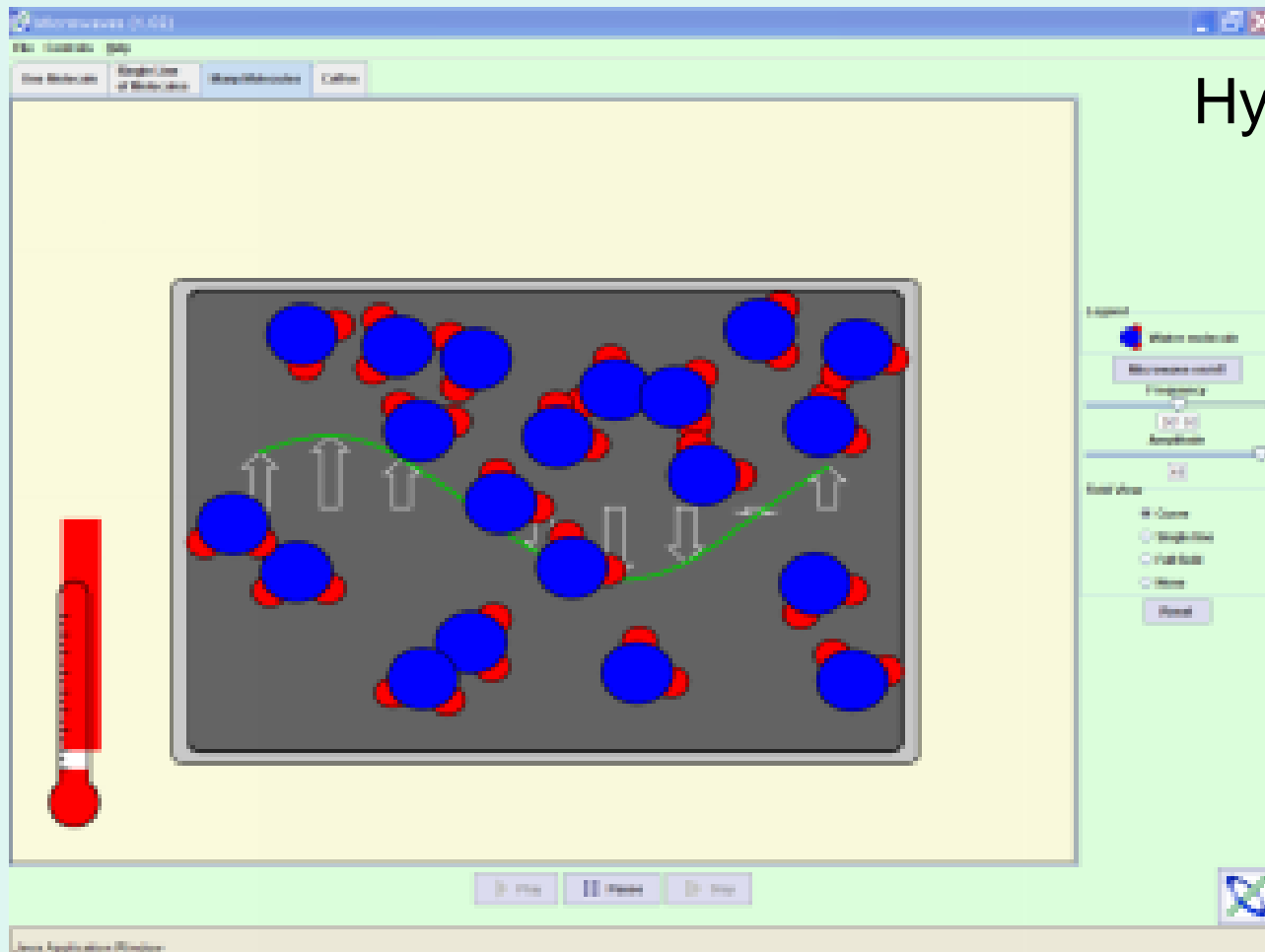


The COMET Program

The resonant frequency of greenhouse gases is in the IR region

Carbon dioxide, water vapour, methane, nitrous oxide, and a few other gases are greenhouse gases. They all are molecules composed of more than two component atoms, bound loosely enough together to be able to vibrate with the absorption of heat. The major components of the atmosphere N_2 and O_2 are two-atom molecules too tightly bound together to vibrate and thus they do not absorb heat and do not contribute to the greenhouse effect.

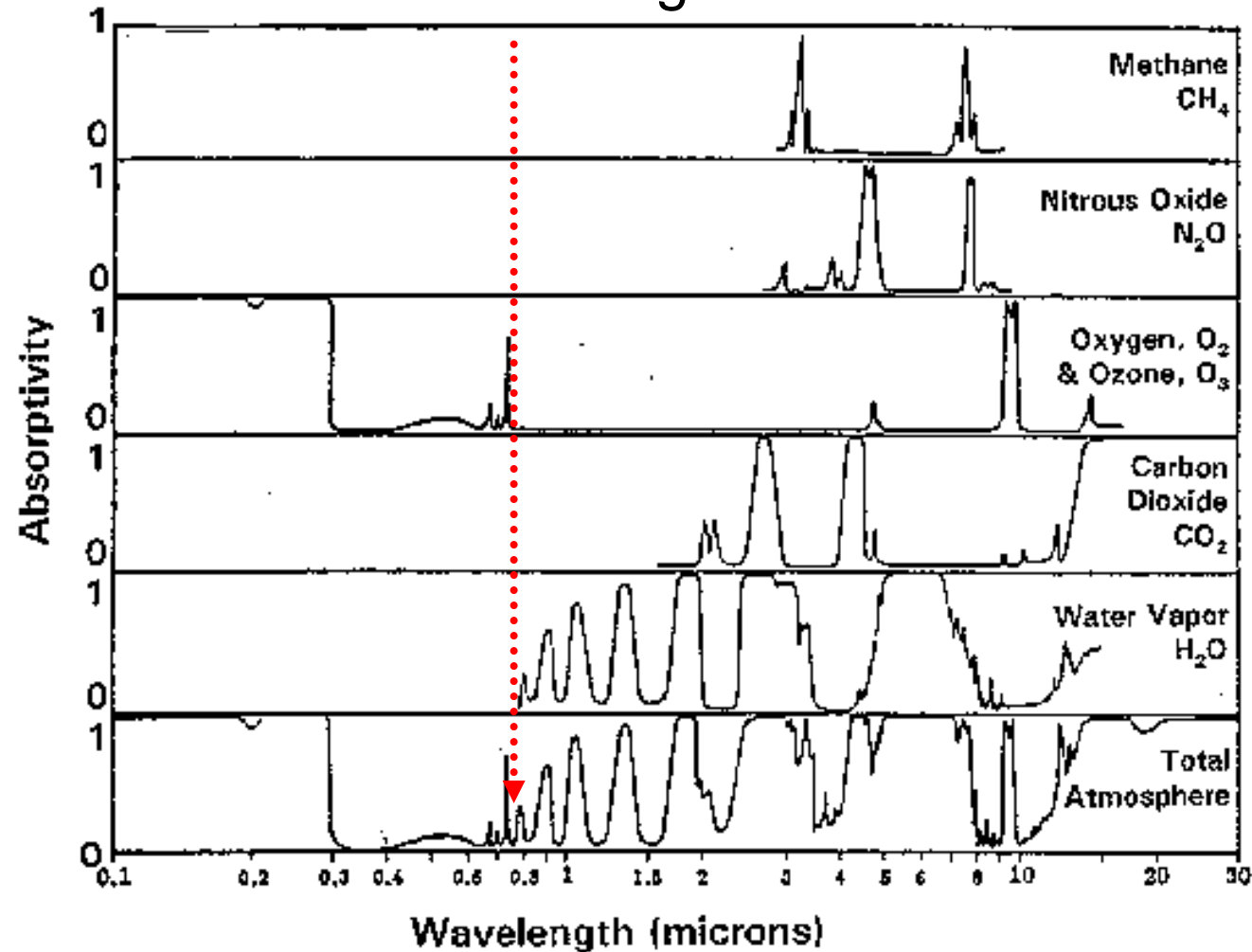
Microwave simulation



Hyperlink

ABSORPTION SPECTRA FOR MAJOR NATURAL GREENHOUSE GASES IN THE EARTH'S ATMOSPHERE

Start of IR region 0.7 μm

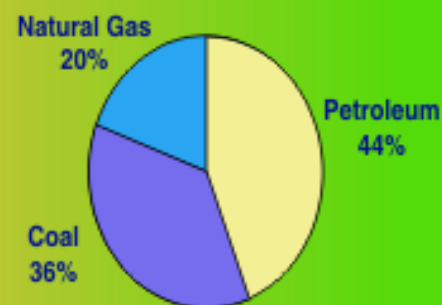
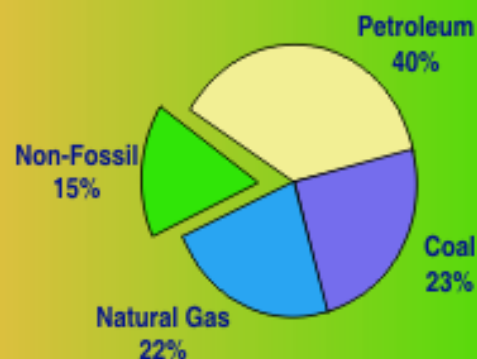


[After J. N. Howard, 1959: *Proc. I.R.E.* 47, 1459; and R. M. Goody and G. D. Robinson, 1951: *Quart. J. Roy. Meteorol. Soc.* 77, 153]

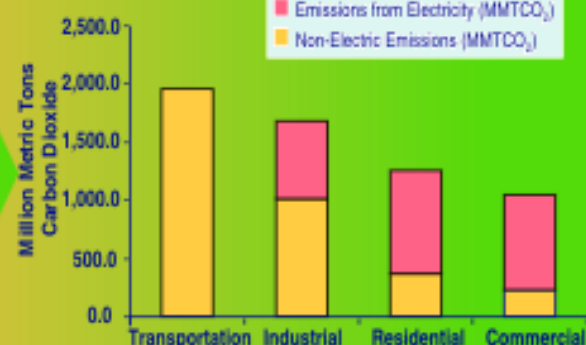
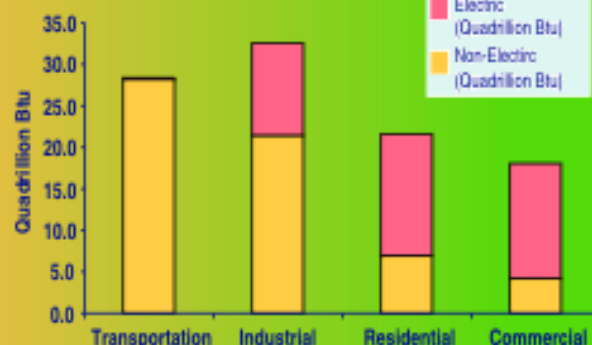
U.S. Primary Energy Consumption

Resulting Carbon Dioxide Emissions

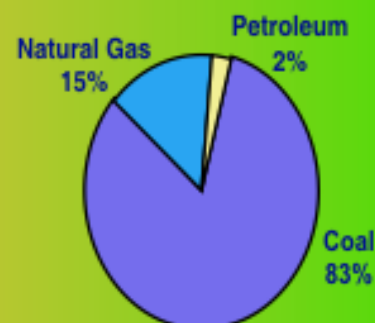
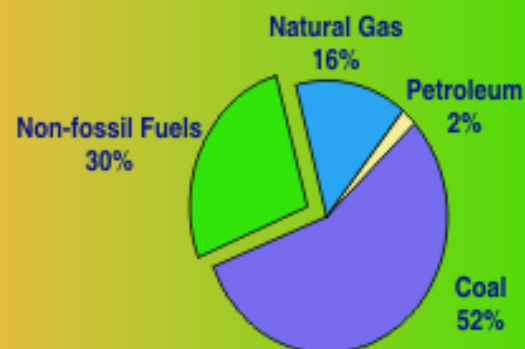
By Fuel Type



By Energy-Use Sector



By Electricity Fuels*



Note: Totals may not equal sum of components due to independent rounding.

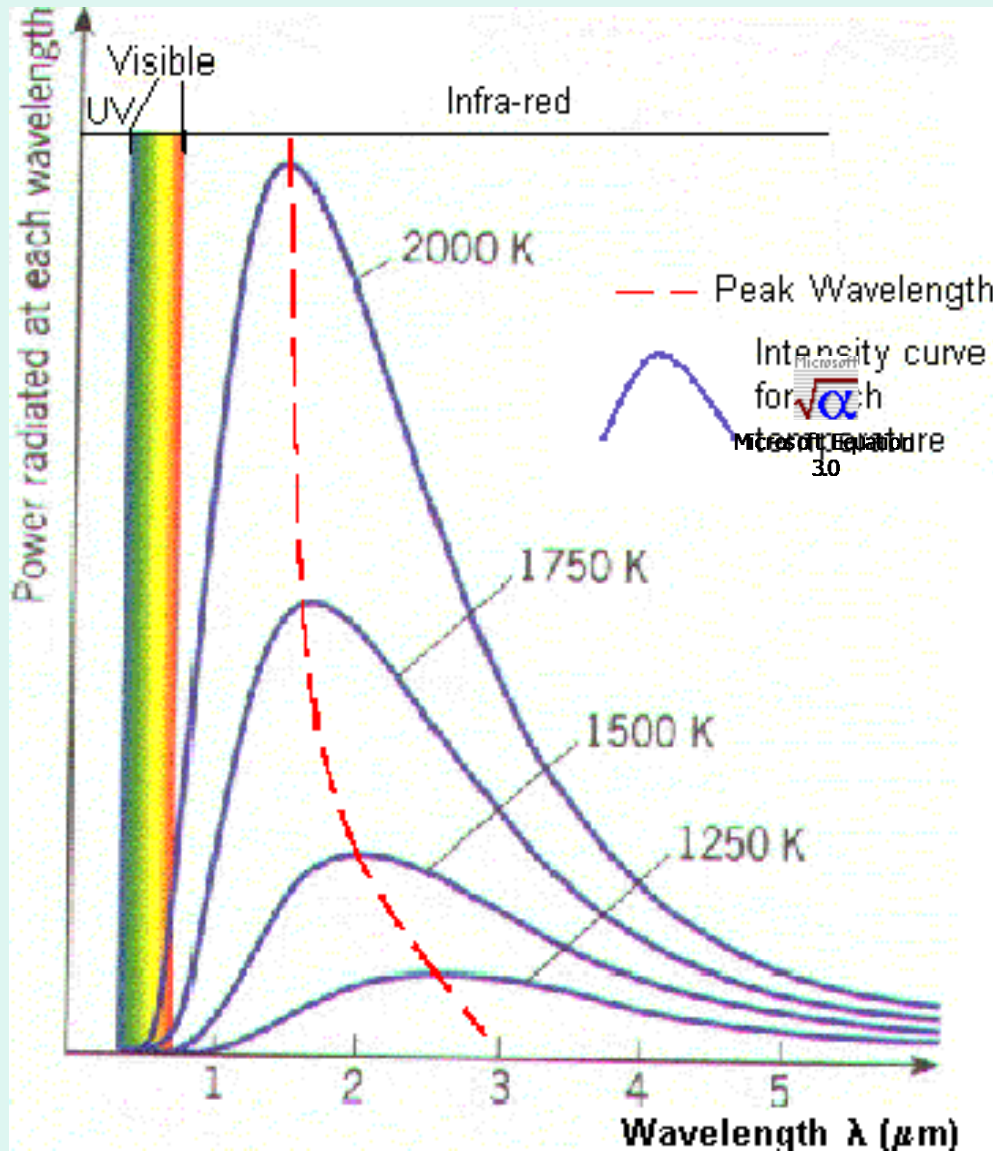
Source: Energy Information Administration. *Electric power sector only.

Sources of greenhouse gases

- Burning of fossil fuels and deforestation leading to higher carbon dioxide concentrations
- Livestock CO_2 and CH_4
- Fertilisers N_2O
- CFC's in refrigeration and fire extinguishers
- When these gases are ranked by their contribution to the greenhouse effect, the most important are:
 - water vapour, which contributes 36–70%
 - carbon dioxide, which contributes 9–26%
 - methane, which contributes 4–9%
 - ozone, which contributes 3–7%

How much heat does the Earth
radiate?

The nature of black-body radiation.



$$\lambda_{\text{max}} \times T = \text{Wien's constant}$$

Stefan–Boltzmann law

$$P = \sigma A T^4$$

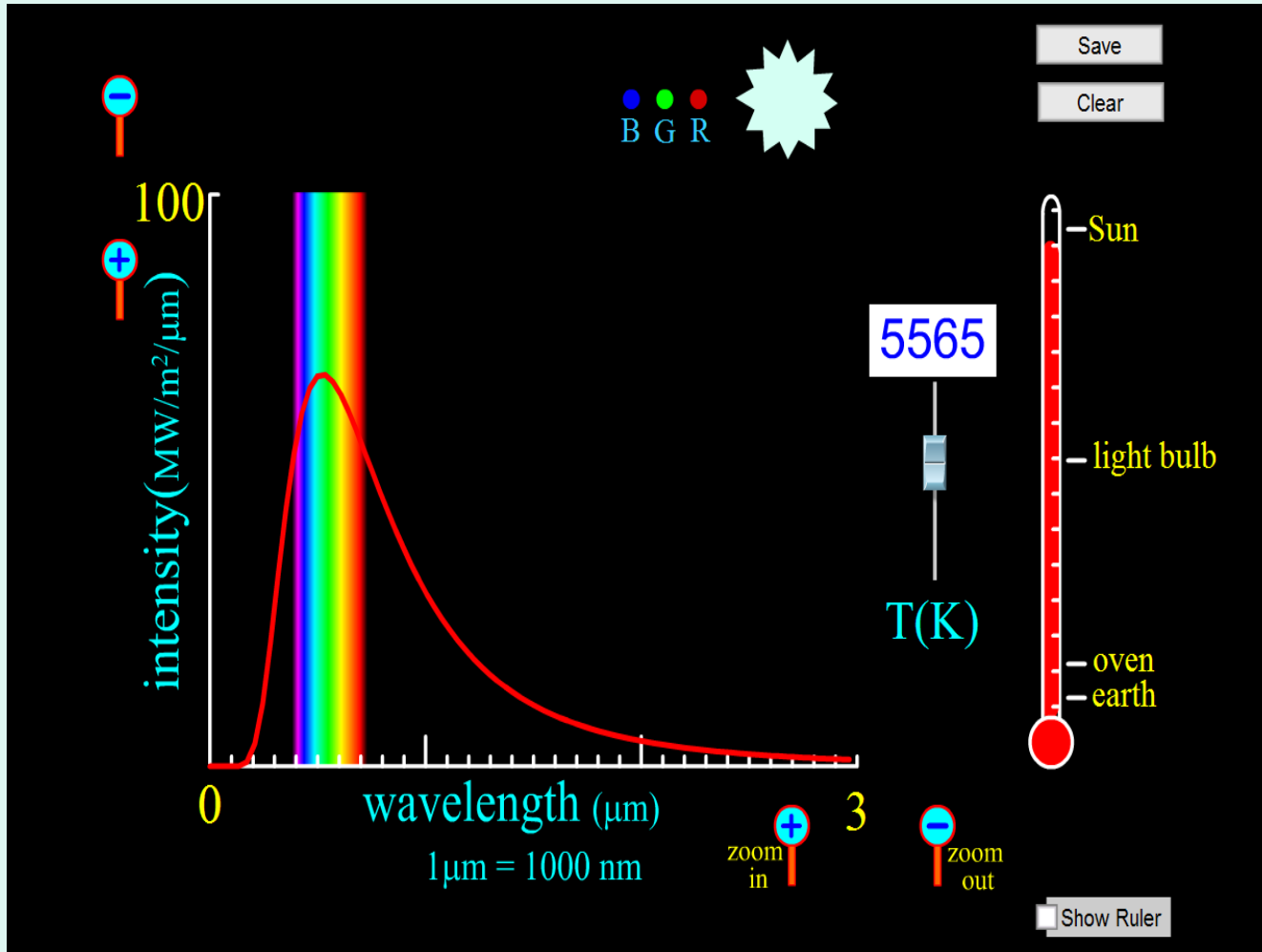
P = Power output

σ = Stefan–Boltzmann constant

A = Surface area of emitting body

T = Temperature of the emitter

Black body simulation



Emissivity

- The Earth is not a perfect Black Body radiator
- The emissivity is defined as

$$\frac{\text{power} - \text{radiated} - \text{by} - \text{object}}{\text{Power} - \text{from} - \text{black} - \text{body} - \text{at} - \text{same} - \text{temp}}$$

Therefore the Earth is not a perfect absorber or emitter of heat. Black objects have a high emissivity, white low.

$$\text{power} = e \sigma A T^4$$

Values of emissivity

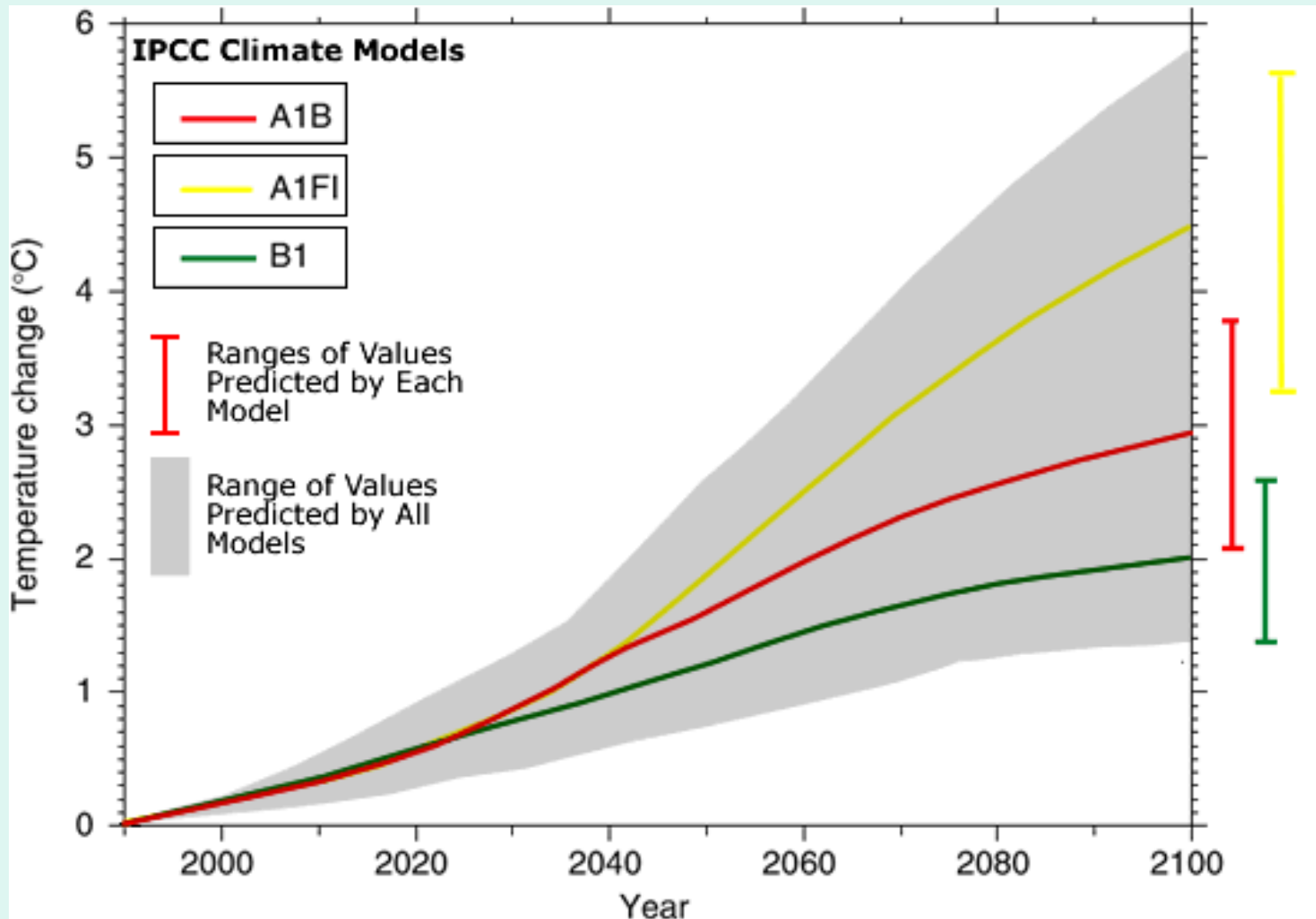
Aluminium: anodised	0.77
Aluminium: polished	0.05
Asbestos: board	0.96
Asbestos: fabric	0.78
Asbestos: paper	0.93
Asbestos: slate	0.96
Brass: highly polished	0.03
Brass: oxidized	0.61
Brick: common	.81-.86
Brick: common, red	0.93
Brick: facing, red	0.92
Brick: fireclay	0.75
Brick: masonry	0.94
Brick: red	0.90
Carbon: candle soot	0.95
Carbon: graphite, filed surface	0.98

What is the effect of the absorbed radiation on the temperature of the Earth?

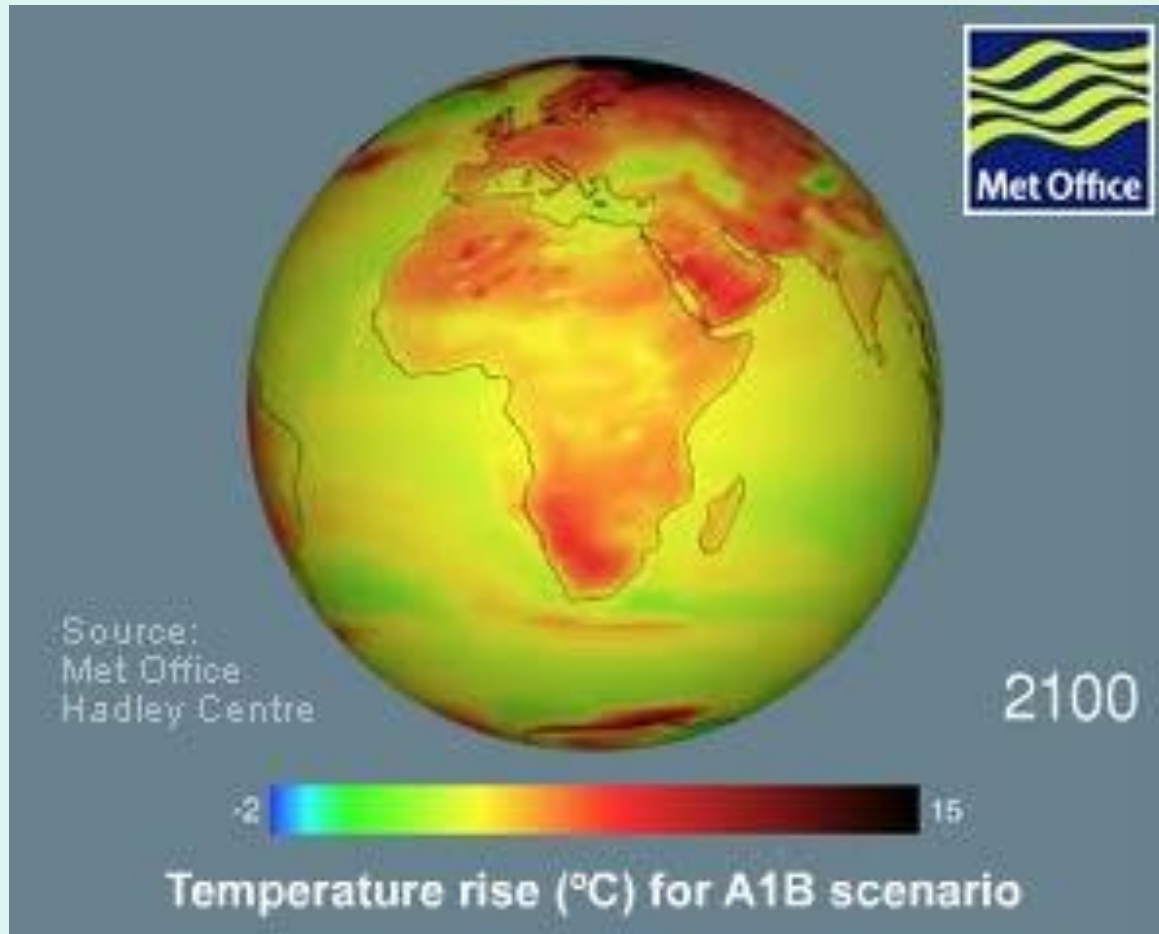
Greenhouse simulation



Predictions



Met office prediction



GLOBAL WARMING

Describe some possible models of global warming.

Students must be aware that a range of models has been suggested to explain global warming, including changes in the composition of **greenhouse gases** in the atmosphere, increased **solar flare activity**, cyclical changes in the **Earth's orbit** and **volcanic activity**.

State what is meant by the enhanced greenhouse effect.

It is sufficient for students to be aware that enhancement of the greenhouse effect is caused by **human activities**.

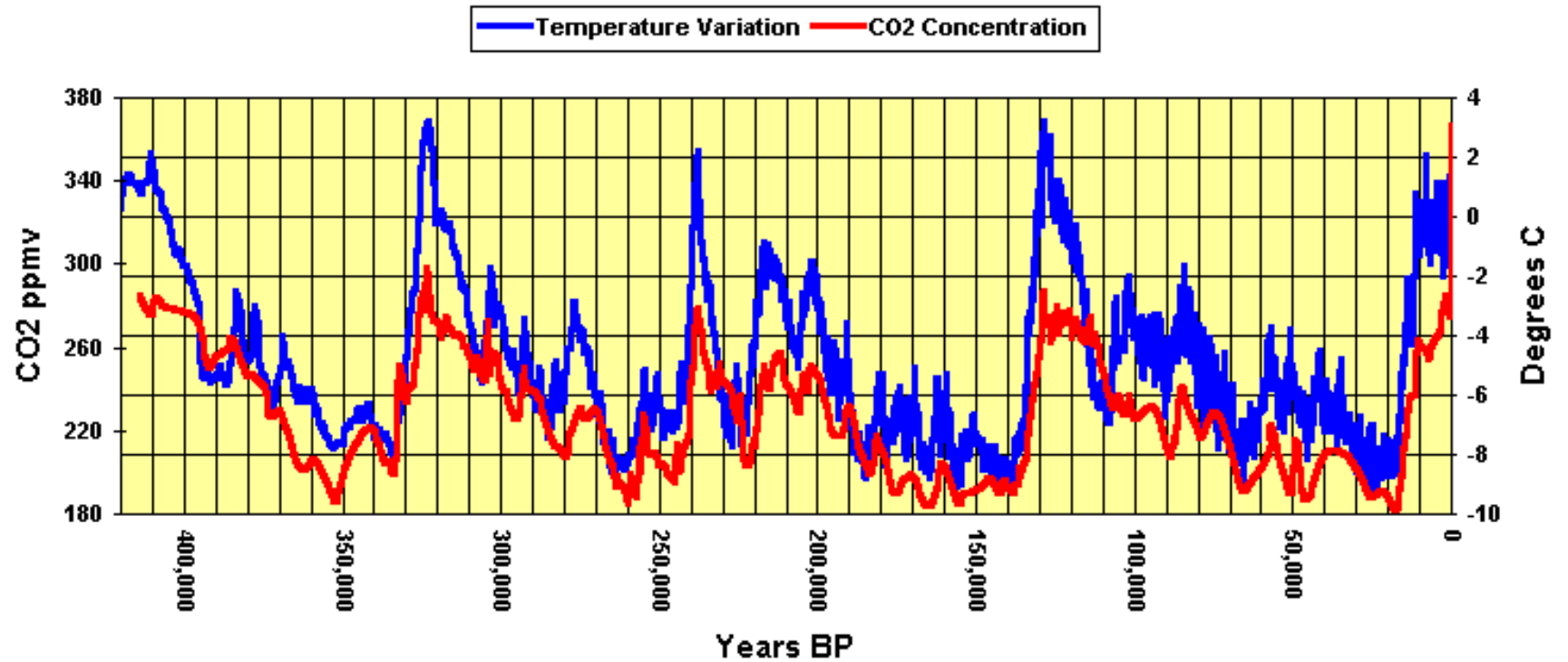
Identify the increased combustion of fossil fuels as the likely major cause of the enhanced greenhouse effect

Students should be aware that, although debatable, the generally accepted view of most scientists is that human activities, mainly related to burning of fossil fuels, have released extra carbon dioxide into the atmosphere.

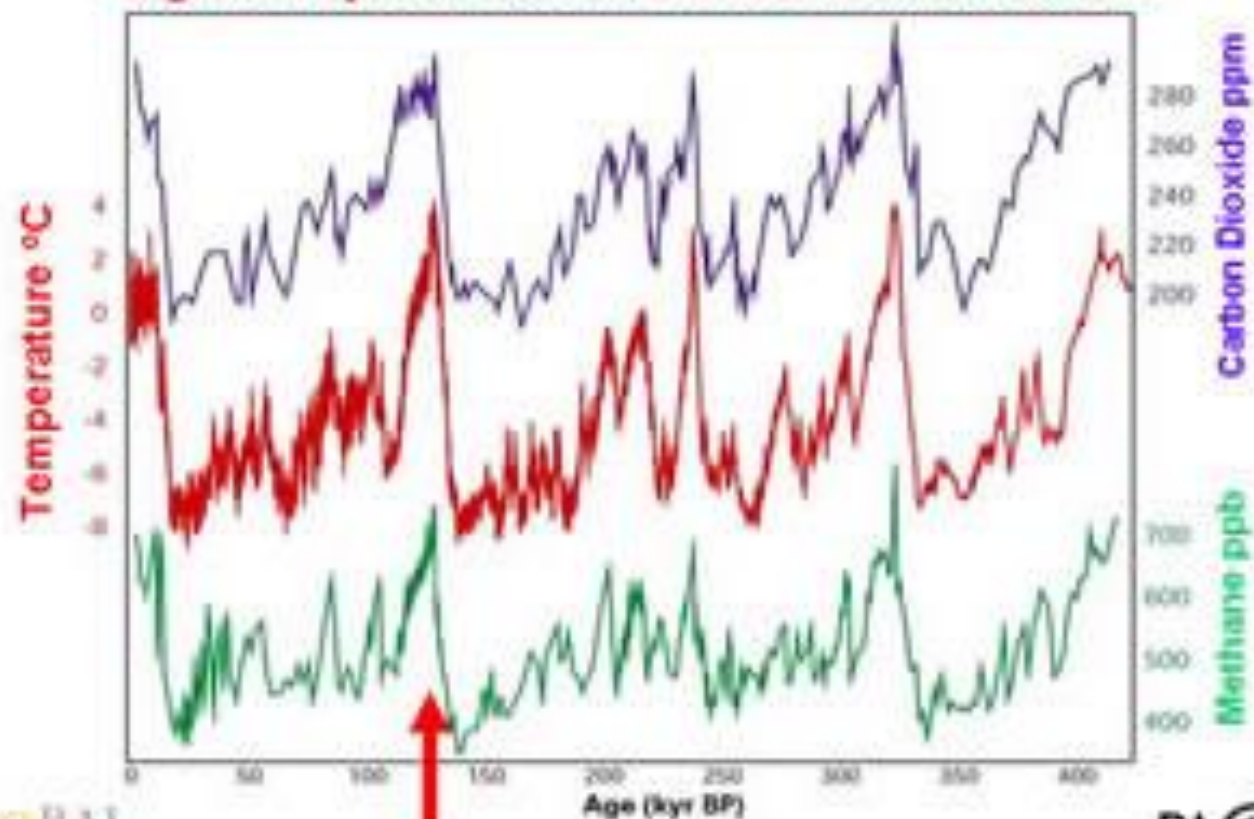
Describe the evidence that links global warming to increased levels of greenhouse gases.

- For example, international ice core research produces evidence of atmospheric composition and mean global temperatures over thousands of years (ice cores up to 420,000 years have been drilled in the Russian Antarctic base, Vostok).

Antarctic Ice Core Data 1



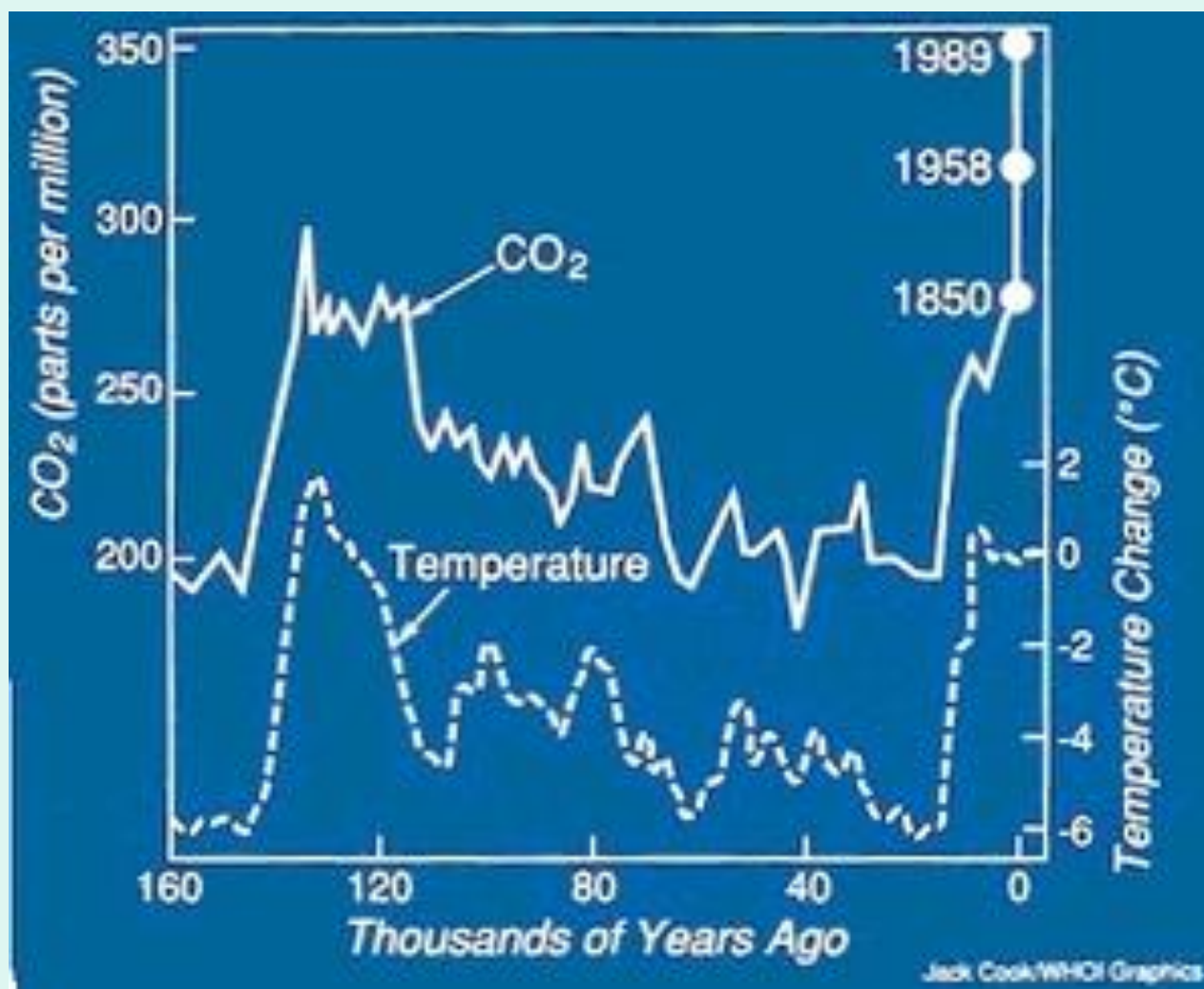
4 glacial cycles recorded in the Vostok ice core

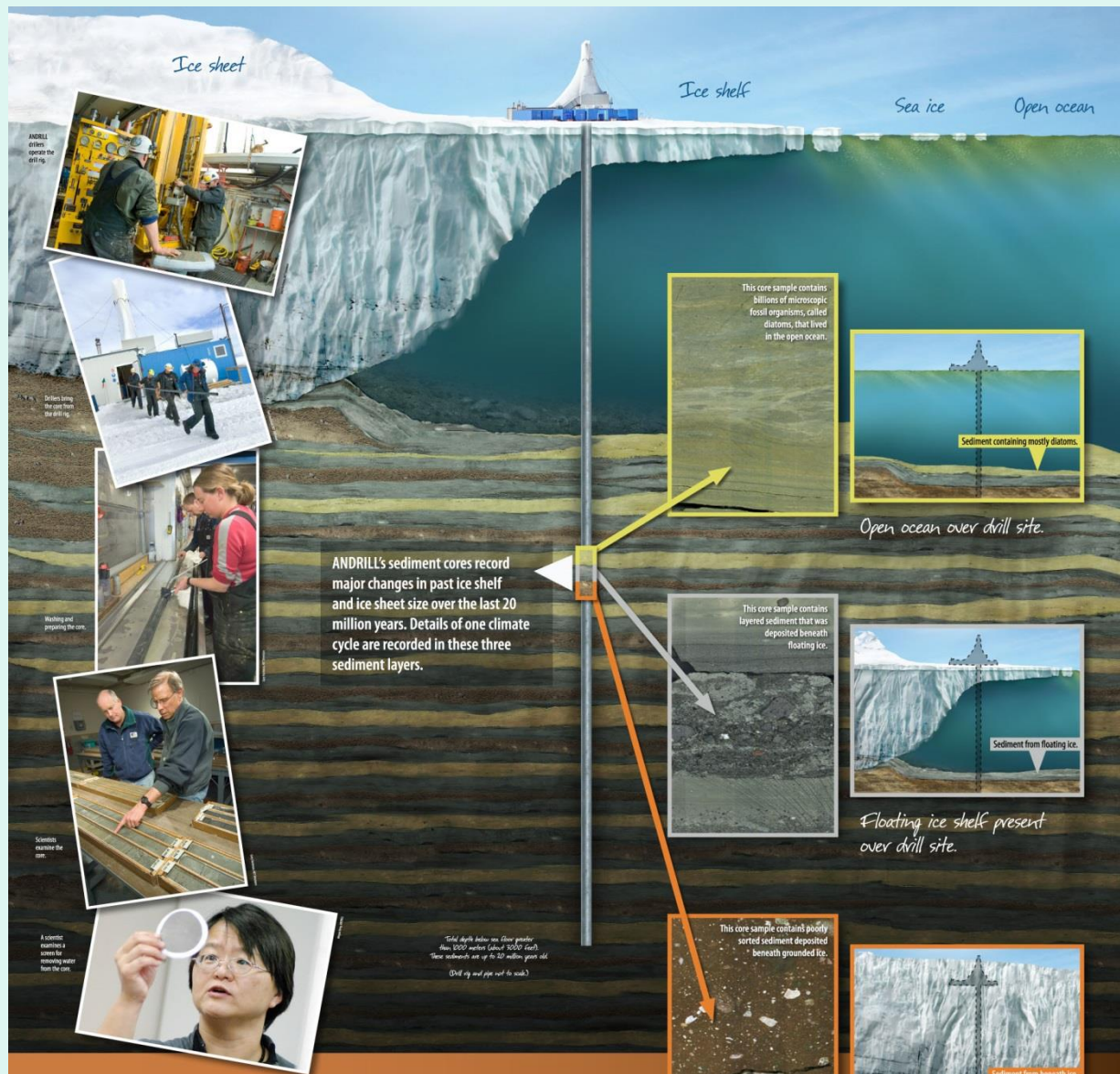


GLOBAL
CHANGE

J.R. Petit et al., *Nature*, 399, 429-36, 1999.

PAGES
PAST GLOBAL CHANGE

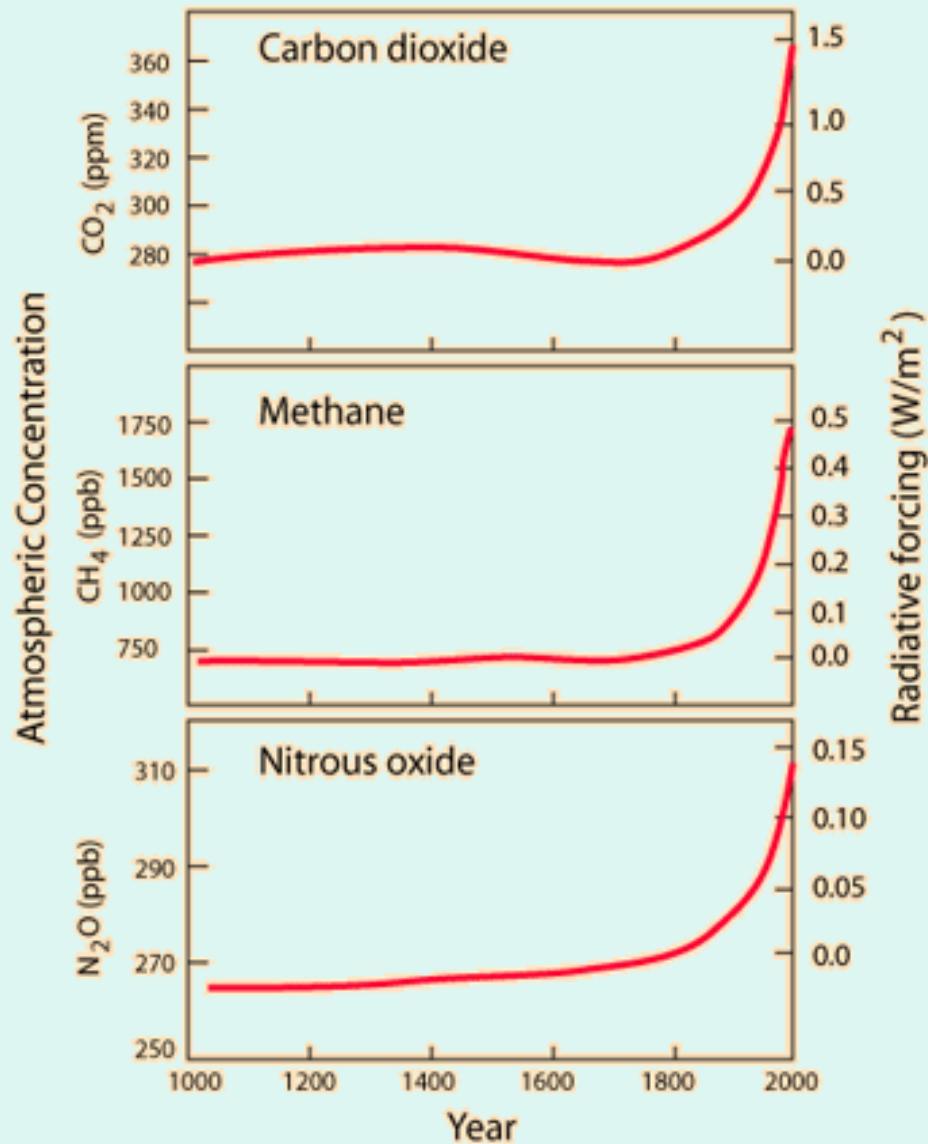




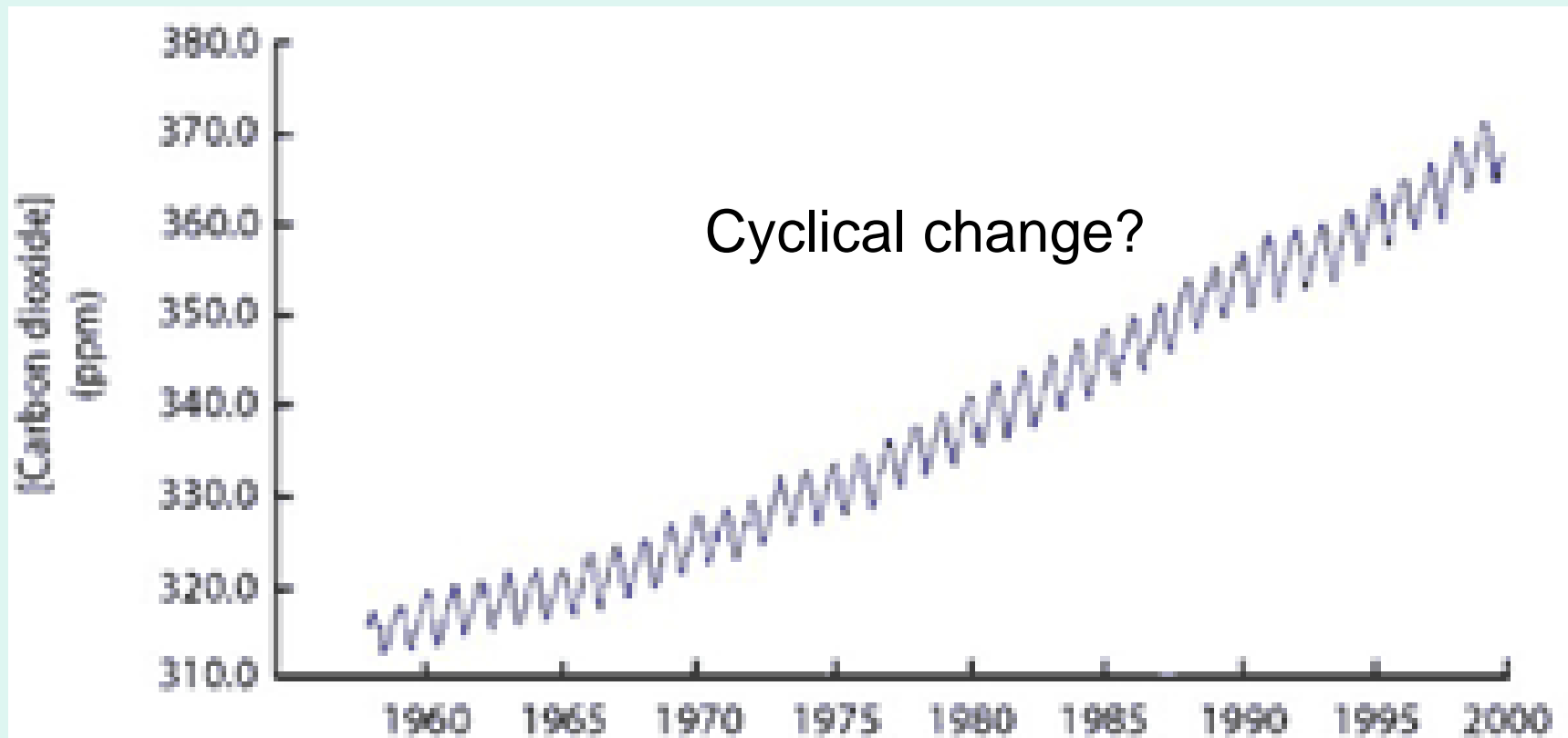
Reading Antarctica's ROCK CORES

ANDRILL is an international program drilling deep into the sediments around Antarctica to better understand the dynamic history and future of Earth's climate. Features of these sediments and enclosed fossils indicate past changes in the Antarctic environment.

Evidence of Global warming



The concentration of carbon dioxide measured at Mauna Loa Observatory in Hawaii



Outline some of the mechanisms that may increase the rate of global warming.

Students should know that:

- global warming reduces ice/snow cover, which in turn changes the albedo, to increase rate of heat absorption
- temperature increase reduces the solubility of CO₂ in the sea and increases atmospheric concentrations
- deforestation reduces carbon fixation.