

Saturated: solution containing all the \_\_\_\_\_ possible at given conditions of \_\_\_\_\_ and \_\_\_\_\_.

Unsaturated: solution containing \_\_\_\_\_ dissolved \_\_\_\_\_ than the \_\_\_\_\_ amount that can be \_\_\_\_\_ at given conditions of \_\_\_\_\_ and \_\_\_\_\_.

Supersaturated: unusual solution containing \_\_\_\_\_ dissolved \_\_\_\_\_ than is normally \_\_\_\_\_ at given conditions of \_\_\_\_\_ and \_\_\_\_\_.

Solubility

A \_\_\_\_\_ of how much \_\_\_\_\_ can \_\_\_\_\_ in a given amount of \_\_\_\_\_ at a specific \_\_\_\_\_.

Dilute Solution: The amount of \_\_\_\_\_ dissolved is \_\_\_\_\_ in relation to the amount of \_\_\_\_\_ present.

Concentrated Solution: The amount of \_\_\_\_\_ dissolved is \_\_\_\_\_ in relation to the amount of \_\_\_\_\_ present.

## Factors Affecting Solubility

1) \_\_\_\_\_ of \_\_\_\_\_ and \_\_\_\_\_:  
" \_\_\_\_\_ dissolves \_\_\_\_\_".

2) \_\_\_\_\_  
□ generally, increasing the \_\_\_\_\_ of the solution  
\_\_\_\_\_ the solubility of a \_\_\_\_\_ solute  
□ increasing the \_\_\_\_\_ of the solution \_\_\_\_\_  
the solubility of a \_\_\_\_\_ solute

3) \_\_\_\_\_  
□ only affects the \_\_\_\_\_ of a \_\_\_\_\_  
\_\_\_\_\_  
□ \_\_\_\_\_ pressure \_\_\_\_\_ solubility  
□ \_\_\_\_\_ pressure \_\_\_\_\_ solubility  
□ Henry's Law: The \_\_\_\_\_ of a \_\_\_\_\_ dissolved  
in a given \_\_\_\_\_ of \_\_\_\_\_ is  
\_\_\_\_\_ proportional to the \_\_\_\_\_ of the  
\_\_\_\_\_.

## The Chemistry Quiz

CR1. \_\_\_\_\_ CR2. \_\_\_\_\_

1. \_\_\_\_\_ 2. \_\_\_\_\_ 3. \_\_\_\_\_ 4. \_\_\_\_\_ 5. \_\_\_\_\_