## **Accuracy in Measurement**

No measurement is 100 percent accurate. All measurements have some degree of uncertainty. When taking measurements, you should always ask yourself, "How accurate is this measurement?" For a measurement to be of any worth, it must have something that indicates its reliability. A measurement's **accuracy** is expressed as its potential amount of error. For instance, the smallest unit of measurement on a metric ruler is usually a millimeter. The most accurate measurement you could possibly make with that ruler is to the nearest millimeter; thus, the measurement's accuracy

is  $\pm$  0.5 mm. This is important because not all measurements have the same accuracy. The total accuracy of your work is only as reliable as your *least* accurate measurement. Following is an example:

Ricardo added the following three liquids to a beaker:

- 7.9 mL of liquid A
- 2.1 mL of liquid B
- 250 mL of liquid C

Ricardo measured liquids *A* and *B* with a narrow graduated cylinder that had markings for every 0.1 mL. He measured liquid *C* in a beaker that had markings only for every 10 mL. Thus, the volume of liquid *C* was only accurate to within about 5 mL. As a result, Ricardo correctly stated that the total volume of the

mixture in the beaker was (7.9 + 2.1 +250) mL =260 mL  $\pm$  5 mL.

## Matchmaker

Match the measurement devices below with their level of accuracy.

- 1. \_\_\_\_\_ metric ruler with markings as small as millimeters
- 2. \_\_\_\_\_ graduated cylinder with markings as small as 2 mL
- 3. \_\_\_\_\_ scale with markings as small as 0.01 g
- 4. \_\_\_\_\_ thermometer with markings as small as 1°C
- a. about 0.5 g
- **b.** about 1 mL
- c. about 0.5°C
- d. about 1 mm
- e. about 0.5 mm
- **f.** about 0.005 g