

## Naming Acids and Bases

*Name the following acids and bases:*

- 1) NaOH \_\_\_\_\_
- 2) H<sub>2</sub>SO<sub>3</sub> \_\_\_\_\_
- 3) H<sub>2</sub>S \_\_\_\_\_
- 4) H<sub>3</sub>PO<sub>4</sub> \_\_\_\_\_
- 5) NH<sub>3</sub> \_\_\_\_\_
- 6) HCN \_\_\_\_\_
- 7) Ca(OH)<sub>2</sub> \_\_\_\_\_
- 8) Fe(OH)<sub>3</sub> \_\_\_\_\_
- 9) H<sub>3</sub>P \_\_\_\_\_

*Write the formulas of the following acids and bases:*

- 10) hydrofluoric acid \_\_\_\_\_
- 11) hydroselenic acid \_\_\_\_\_
- 12) carbonic acid \_\_\_\_\_
- 13) lithium hydroxide \_\_\_\_\_
- 14) nitrous acid \_\_\_\_\_
- 15) cobalt (II) hydroxide \_\_\_\_\_
- 16) sulfuric acid \_\_\_\_\_
- 17) beryllium hydroxide \_\_\_\_\_
- 18) hydrobromic acid \_\_\_\_\_