

IB/HL2 Chemistry

Name _____

Formula Writing Exercises

Date _____

Write formulas for the following

_____ 1. Copper (II) Chloride

_____ 2. Cesium Bromide

_____ 3. Potassium nitrate

_____ 4. Silver Chromate

_____ 5. Zinc Hydroxide

_____ 6. Iron (II) Carbonate

_____ 7. Iron (III) Carbonate

_____ 8. Magnesium Nitrate

_____ 9. Barium sulfate

_____ 10. Copper (I) Oxide

_____ 11. Copper (II) Oxide

_____ 12. Calcium Chloride

_____ 13. Iron (II) Phosphate

_____ 14. Aluminum Phosphate

_____ 15. Sodium Sulfate

_____ 16. Sodium Oxalate

_____ 17. Strontium Nitrate

_____ 18. Copper (II) Hydroxide

_____ 19. Iron (II) Sulfate

_____ 20. Iron (III) Chloride

_____ 21. Cobalt Chloride

_____ 22. Lithium Chloride

_____ 23. Lithium Carbonate

_____ 24. Tin (IV) iodide

_____ 25. Lead (II) Oxide

Write names for the following Compounds

1 KOH

2 CuSO₄

3 AlBr₃

4 (NH₄)₂SO₄

5 KBr

6 KCl

7 KClO₃

- 8 NaClO
- 9 BaCl_2
- 10 ZnO
- 11 AgNO_3
- 12 Ag_2CrO_4
- 13 KMnO_4
- 14 $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$
- 15 CsCl
- 16 FeS
- 17 NaI
- 18 K_3PO_4
- 19 $\text{NaC}_2\text{H}_3\text{O}_2$
- 20 PbCl_2
- 21 SnO
- 22 SnO_2
- 23 PbO
- 24 NaF
- 25 MgSO_4
- 26 ZnSO_3
- 27 HgCl_2
- 28 Hg_2Cl_2
- 29 Al_2O_3
- 30 $\text{Sr}_3(\text{PO}_4)_2$

