## DO NOT WRITE ON THIS PAPER!

# **Holt Physical Science: Section 4.1 Atomic Structure**

### What are Atoms?

- 1. What did Democritus suggest?
- 2. The word atom is derived from the Greek word meaning "\_\_\_\_\_.".
- 3. What did John Dalton propose in 1808 (be specific)? Why was it widely accepted?

### What's in an Atom?

- 4. What three subatomic particles are involved in the everyday chemistry of most substances? (Include a detailed description of each.)
- 5. Even though atoms are made of charged particles, why don't atoms have a charge? Give an example.

### Models of the Atom.

- 6. What happened to the original theories about the structure of the atom? Explain.
- 7. Describe Bohr's atomic model. (Be specific)
- 8. What does the modern theory of the atom (specifically) state?
- 9. How do scientists determine an electron's position? How do they show it?
- 10. What is the area where an electron is found called? How many are there, what are their names, and what do they look like?

What is a valence electron?