Global Science A Final Study Guide

Chemistry

- 1. What is in an atomic nucleus and what are their charges?
- 2. What is the charge of an electron?
- 3. What determines an atom's mass?
- 4. Describe the characteristics of alkali metals. Where are they found on the periodic table?
- 5. What is a covalent bond?
- 6. What is an ionic bond?
- 7. Which electron level is involved when atoms bond?
- 8. Which electron level is used in a Lewis dot diagram?
- 9. Is Na⁺ an anion or a cation? How many electrons did Na lose or gain to be an ion? How can you tell?
- 10. What is the order of the periodic table based on?
- 11. Using the periodic table, know how to find the atomic number, number of protons, number of electrons, number of neutrons, and atomic mass of an element.
- 12. What is an isotope?
- 13. What makes noble gases inert?

Waves

- 1. What do waves transport?
- 2. What is the crest, trough, amplitude, and wavelength of a wave? Be able to identify these parts in a diagram.
- 3. In which direction do particles move in a transverse wave?
- 4. What do seismic waves travel through?
- 5. What type of waves require a medium?
- 6. What is reflection, refraction, diffraction, and interference?

Biodiversity

- 1. Define species, population, community, and ecosystem?
- 2. What is biodiversity?
- 3. What is a decomposer?
- 4. What is a scavenger?
- 5. How can the introduction of an exotic species change an ecosystem?
- 6. What is carrying capacity?

Biogeochemical Cycles

- 1. Define the following: evaporation, condensation, precipitation, and transpiration.
- 2. What is the chemical equation for photosynthesis (what goes in and what comes out)?
- 3. What is the chemical equation for respiration?
- 4. How is carbon cycled through ecosystems?
- 5. Where does photosynthesis occur? How do plants get the carbon they need to make sugar?

<u>I & E</u>

Be able to interpret graphs and tables.